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## Structure of a 2,2'-Bipyridine Containing Dioxolane and 6-Bromopyridine Subunits

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**Abstract.** 6,6'-Bis[2-(6-bromo-2-pyridyl)-1,3-dioxolan-2-yl]-2,2'-bipyridine,  $C_{26}H_{20}Br_2N_4O_4$ ,  $M_r = 612.3$ , triclinic,  $P\bar{1}$ ,  $a = 6.231(1)$ ,  $b = 9.859(2)$ ,  $c = 10.299(3)$  Å,  $\alpha = 88.45(2)$ ,  $\beta = 84.65(2)$ ,  $\gamma = 77.09(2)^\circ$ ,  $V = 613.9(3)$  Å<sup>3</sup>,  $Z = 1$ ,  $D_x = 1.656$  g cm<sup>-3</sup>, Mo  $K\alpha$ ,  $\lambda = 0.71073$  Å,  $\mu = 33.1$  cm<sup>-1</sup>,  $F(000) = 306$ ,  $T = 302$  K,  $R = 0.037$  for 1795 observations (of 2158 unique data). The molecule has crystallographic  $\bar{1}$  symmetry, with the bipyridine moiety in the *anti* conformation, the dioxolane rings in the half-chair conformation, and pyridine nitrogen atoms oriented *anti* to dioxolane oxygen atoms.

**Experimental.** Material synthesized by Taylor (1983), colorless crystals from CH<sub>3</sub>CN. Crystal size 0.28 × 0.32 × 0.48 mm, space group by successful refinement of centrosymmetric model; cell dimensions from setting angles of 25 reflections having  $14 < \theta < 16^\circ$ . Data collection on Enraf-Nonius CAD-4 diffractometer, Mo  $K\alpha$  radiation, graphite monochromator,  $\omega-2\theta$  scans designed for  $I = 25\sigma(I)$ . Scan rates varied 0.49–5.0° min<sup>-1</sup>. Reflections having  $1 < \theta < 25^\circ$ ,  $0 \leq h \leq 7$ ,  $-11 \leq k \leq 11$ ,  $-12 \leq l \leq 12$  measured, corrected for background, Lorentz and polarization effects. No redundant data measured. Absorption corrections by  $\psi$  scans of reflections near  $\chi = 90^\circ$ , minimum relative transmission coefficient 69.4%. Standard reflections 300, 030, 006,  $\pm 2.1\%$  random variation. Structure solved by heavy-atom methods, refined by full-matrix least squares based on  $F$ , using data for which  $I > 3\sigma(I)$ , weights  $w = \sigma^{-2}(F_o)$ , with Enraf-Nonius

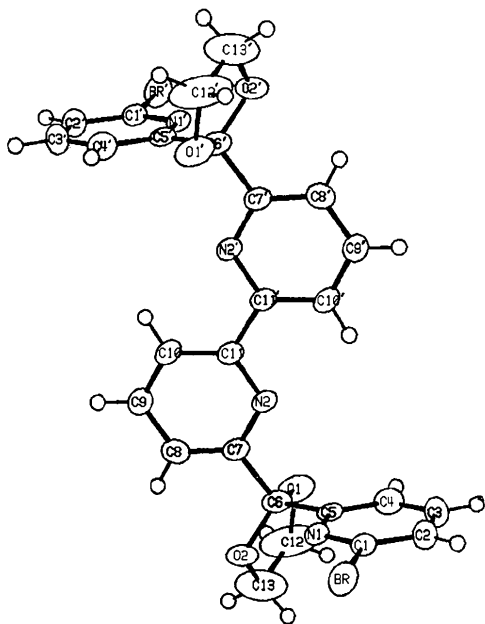


Fig. 1. Numbering scheme and conformation.

Table 1. Coordinates and equivalent isotropic thermal parameters

	x	y	z	$B_{eq}^*(\text{Å}^2)$
Br	0.15335 (6)	0.32311 (3)	0.06789 (3)	3.800 (7)
O(1)	0.1375 (5)	0.8394 (2)	0.4299 (2)	4.31 (6)
O(2)	-0.1155 (4)	0.7050 (2)	0.4298 (2)	3.67 (5)
N(1)	0.1299 (4)	0.5556 (2)	0.2148 (2)	2.42 (5)
N(2)	0.0327 (4)	0.8917 (2)	0.1445 (2)	2.60 (5)
C(1)	0.2741 (5)	0.4529 (3)	0.1559 (3)	2.62 (6)
C(2)	0.5012 (5)	0.4301 (4)	0.1551 (4)	3.61 (8)
C(3)	0.5823 (5)	0.5237 (4)	0.2220 (4)	4.21 (8)
C(4)	0.4352 (6)	0.6323 (3)	0.2858 (4)	3.81 (7)
C(5)	0.2111 (5)	0.6462 (3)	0.2801 (3)	2.53 (6)
C(6)	0.0382 (5)	0.7623 (3)	0.3472 (3)	2.94 (6)
C(7)	-0.0875 (5)	0.8641 (3)	0.2521 (3)	2.66 (6)
C(8)	-0.3075 (6)	0.9261 (3)	0.2800 (3)	3.45 (7)
C(9)	-0.4124 (6)	1.0200 (4)	0.1897 (4)	3.64 (8)
C(10)	-0.2915 (6)	1.0506 (3)	0.0795 (3)	3.20 (7)
C(11)	-0.0692 (5)	0.9855 (3)	0.0591 (3)	2.55 (6)
C(12)	0.0974 (12)	0.7937 (6)	0.5585 (4)	8.8 (2)
C(13)	-0.0469 (11)	0.6965 (6)	0.5552 (4)	7.7 (2)

$$* B_{eq} = \frac{1}{3}(a^2\beta_{11} + b^2\beta_{22} + c^2\beta_{33} + ab\beta_{12}\cos\gamma + ac\beta_{13}\cos\beta + bc\beta_{23}\cos\alpha).$$

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Table 2. Bond distances (Å), angles (°) and torsion angles (°)

Br—C(1)	1.909 (3)	C(3)—C(4)	1.380 (5)
O(1)—C(6)	1.425 (3)	C(4)—C(5)	1.379 (4)
O(1)—C(12)	1.405 (5)	C(5)—C(6)	1.517 (4)
O(2)—C(6)	1.422 (3)	C(6)—C(7)	1.522 (3)
O(2)—C(13)	1.394 (5)	C(7)—C(8)	1.378 (4)
N(1)—C(1)	1.314 (3)	C(8)—C(9)	1.394 (4)
N(1)—C(5)	1.343 (3)	C(9)—C(10)	1.368 (4)
N(2)—C(7)	1.336 (3)	C(10)—C(11)	1.390 (4)
N(2)—C(11)	1.352 (3)	C(11)—C(11')	1.484 (4)
C(1)—C(2)	1.381 (4)	C(12)—C(13)	1.456 (7)
C(2)—C(3)	1.376 (5)		
C(6)—O(1)—C(12)	107.4 (3)	O(1)—C(6)—C(7)	107.9 (2)
C(6)—O(2)—C(13)	107.5 (3)	O(2)—C(6)—C(5)	109.8 (2)
C(1)—N(1)—C(5)	116.9 (2)	O(2)—C(6)—C(7)	108.9 (2)
C(7)—N(2)—C(11)	117.6 (2)	C(5)—C(6)—C(7)	113.3 (2)
Br—C(1)—N(1)	115.9 (2)	N(2)—C(7)—C(6)	115.3 (2)
Br—C(1)—C(2)	118.3 (2)	N(2)—C(7)—C(8)	123.4 (3)
N(1)—C(1)—C(2)	125.8 (3)	C(6)—C(7)—C(8)	121.3 (3)
C(1)—C(2)—C(3)	116.7 (3)	C(7)—C(8)—C(9)	118.5 (3)
C(2)—C(3)—C(4)	118.9 (3)	C(8)—C(9)—C(10)	118.9 (3)
C(3)—C(4)—C(5)	119.8 (3)	C(9)—C(10)—C(11)	119.3 (3)
N(1)—C(5)—C(4)	121.8 (3)	N(2)—C(11)—C(10)	122.2 (3)
N(1)—C(5)—C(6)	114.9 (2)	N(2)—C(11)—C(11')	116.3 (3)
C(4)—C(5)—C(6)	123.3 (3)	C(10)—C(11)—C(11')	121.5 (3)
O(1)—C(6)—O(2)	106.1 (2)	O(1)—C(12)—C(13)	107.4 (4)
O(1)—C(6)—C(5)	110.6 (2)	O(2)—C(13)—C(12)	106.1 (3)
N(1)—C(5)—C(6)—O(1)	170.9 (4)		
N(2)—C(7)—C(6)—O(2)	-159.5 (4)		
O(1)—C(12)—C(13)—O(2)	-9.3 (6)		
C(12)—C(13)—O(2)—C(6)	20.5 (5)		
C(13)—O(2)—C(6)—O(1)	-24.0 (5)		
O(2)—C(6)—O(1)—C(12)	17.9 (4)		
C(6)—O(1)—C(12)—C(13)	-5.5 (6)		

*SDP* (Frenz & Okaya, 1980), scattering factors of Cromer & Waber (1974). Non-H atoms anisotropic; H atoms located by  $\Delta F$  synthesis, placed in calculated positions with C—H 0.95 Å,  $B = 5.0 \text{ \AA}^2$ . Final  $R = 0.037$  (0.053 for all data),  $wR = 0.049$ ,  $S = 1.563$

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## Structure of Tetraphenylphosphonium Bromide

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**Abstract.**  $\text{C}_{24}\text{H}_{20}\text{P}^+\text{Br}^-$ ,  $M_r = 419.3$ , tetragonal,  $\bar{4}$ ,  $a = 11.960$  (2),  $c = 6.967$  (2) Å,  $V = 996.6$  (3) Å<sup>3</sup>,  $Z = 2$ ,  $D_m = 1.45$ ,  $D_x = 1.40 \text{ g cm}^{-3}$ ,  $\lambda(\text{Mo K}\alpha) = 0.71069$  Å,  $\mu = 21.24 \text{ cm}^{-1}$ ,  $F(000) = 428$ , ambient temperature.  $R = 0.066$  for 1006 observed reflections. The cation has  $\bar{4}$  symmetry with P—C bond lengths of 1.800 (6) Å. The dimensions of the phenyl rings agree with standard values.

for 163 variables. Max. shift  $0.01\sigma$  in final cycle, largest residual density  $0.62 \text{ e \AA}^{-3}$ , min.  $-0.45 \text{ e \AA}^{-3}$ . The molecule is depicted in Fig. 1, coordinates are given in Table 1, bond distances, angles and torsion angles are listed in Table 2.\*

**Related literature.** Chisholm, Huffman, Rothwell, Bradley, Kress & Woodruff (1981); James & Williams (1973); Kvick (1976); Newkome, Taylor, Fronczek & Delord (1984).

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\* Lists of H-atom coordinates, anisotropic thermal parameters and structure factor amplitudes have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 42311 (13 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

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